

Board – CBSE

Class – 7

Topic – Heat

1. State similarities and differences between the laboratory thermometer and the clinical thermometer.

**Ans. Similarities.**

- (i) Both clinical and laboratory thermometers have long, narrow, uniform glass tubes.
- (ii) The bulbs of both the thermometers have mercury in them.

**Differences.**

- (i) The temperature range of clinical thermometers is from  $35^{\circ}\text{C}$  to  $42^{\circ}\text{C}$  while that of laboratory thermometer is from  $-10^{\circ}\text{C}$  to  $110^{\circ}\text{C}$ .
- (ii) Clinical thermometer is used to measure the temperature of a human body. However, a laboratory thermometer cannot be used to measure the temperature of a human body.
- (iii) The least count of both the thermometers differs.
- (iv) Unlike a clinical thermometer that can be tilted, a laboratory thermometer is kept upright while reading the temperature values.

2. Give two examples each of conductors and insulators of heat.

**Ans.** Two examples of conductors of heat are.

- (i) Aluminium
- (ii) Iron

Two examples of insulators of heat are.

- (i) Wood
- (ii) Plastic

3. Fill in the blanks.

- (a) The hotness of an object is determined by its \_\_ \_\_.
- (b) Temperature of boiling water cannot be measured by a \_\_ \_\_ thermometer.
- (c) Temperature is measured in degree \_\_ \_\_.
- (d) No medium is required for transfer of heat by the process of \_\_ \_\_.
- (e) A cold steel spoon is dipped in a cup of hot milk. It transfers heat to its other end by the process of \_\_.

(f) Clothes of \_\_\_ colours absorb heat better than clothes of light colours.

- Ans.** (a) The hotness of an object is determined by its **temperature**.
- (b) Temperature of boiling water cannot be measured by a **clinical** thermometer.
- (c) Temperature is measured in degree **Celsius**.
- (d) No medium is required for heat transfer by the process of **radiation**.
- (e) A cold steel spoon is dipped in a cup of hot milk. It transfers heat to its other end by the process of **conduction**.
- (f) Clothes of **dark** colours absorb heat better than clothes of light colours.

**4.** Match the following.

(i)	The land breeze blows during	(a)	summer
(ii)	The sea breeze blows during	(b)	winter
(iii)	Dark coloured clothes are preferred during	(c)	day
(iv)	Light coloured clothes are preferred during	(d)	night

**Ans.**

(i)	The land breeze blows during	(d)	night
(ii)	The sea breeze blows during	(c)	day
(iii)	Dark coloured clothes are preferred during	(b)	winter
(iv)	Light coloured clothes are preferred during	(a)	summer

**5.** Discuss why wearing more layers of clothing during winters keeps us warmer than wearing just one thick piece of clothing.

**Ans.** We prefer wearing more layers of clothing than just one thick piece of clothing during winters because air gets trapped in between the various clothing layers. Being a poor conductor of heat, air prevents heat loss from our body. Hence, layers of clothing keep us warmer than a single layer.

6. Look at Figure. Mark where the heat is being transferred by conduction, convection and radiation.



- Ans.** (i) Transfer of heat from the burner to the pan is by radiation.  
(ii) Transfer of heat from the pan to water is by conduction.  
(iii) Transfer of heat within the water is by convection.

7. In places of hot climate, it is advised that the outer walls of houses be painted white. Explain.

**Ans.** In places of hot climate, it is advised to paint the outer walls of houses white because a light colour such as white reflects most of the heat that falls on it. Hence, a light colour tends to keep the house cool.

8. One litre of water at  $30^{\circ}\text{C}$  is mixed with one litre of water at  $50^{\circ}\text{C}$ . The temperature of the mixture will be

- (a)  $80^{\circ}\text{C}$  (b) more than  $50^{\circ}\text{C}$  but less than  $80^{\circ}\text{C}$   
(c)  $20^{\circ}\text{C}$  (d) between  $30^{\circ}\text{C}$  and  $50^{\circ}\text{C}$

**Ans.** (d) The temperature of the mixture will be between  $30^{\circ}\text{C}$  and  $50^{\circ}\text{C}$ .

9. An iron ball at  $40^{\circ}\text{C}$  is dropped in a mug containing water at  $40^{\circ}\text{C}$ .

The heat will

- (a) Flow from iron ball to water.  
(b) Not flow from iron ball to water or from water to iron ball.  
(c) Flow from water to iron ball.  
(d) Increase the temperature of both.

**Ans.** (b) The heat will not flow from iron ball to water or water to iron ball as both substances have the same temperature.

**10.** A wooden spoon is dipped in a cup of ice cream. Its other end

(a) becomes cold by the process of conduction.

(b) becomes cold by the process of convection.

(c) becomes cold by the process of radiation.

(d) does not become cold.

**Ans.** (d) Its other end does not become cold as wood is a bad conductor of heat.

**11.** Stainless steel pans are usually provided with copper bottoms. The reason for this could be that

(a) Copper bottom makes the pan more durable.

(b) Such pans appear colourful.

(c) Copper is a better conductor of heat than stainless steel.

(d) Copper is easier to clean than stainless steel.

**Ans.** (c) The reason for this is that copper is a better conductor of heat than stainless steel.